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Atomic motions in the $\alpha\beta$ -region of glass-forming polymers: molecular versus mode coupling theory approach

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Abstract

We present fully atomistic molecular dynamics simulation results on a main-chain polymer, 1,4-polybutadiene, in the merging region of the α - and β -relaxations. A real-space analysis reveals the occurrence of localized motions (' β -like') in addition to the diffusive structural relaxation. A molecular approach provides a direct connection between the local conformational changes reflected in the atomic motions and the secondary relaxations in this polymer. Such local processes occur just in the time window where the β -process of the mode coupling theory is expected. We show that the application of this theory is still possible and yields an unusually large value of the exponent parameter. This result might originate from the competition between two mechanisms for dynamic arrest: intermolecular packing and intramolecular barriers for local conformational changes (' β -like').

(Some figures in this article are in colour only in the electronic version)

1. Introduction

Since they do not easily crystallize, polymers are probably the most extensively studied systems in relation to the glass-transition phenomenon and its associated structural relaxation. They are thus implicitly considered as 'standard' glass-forming systems. However, the macromolecular character of their structural units must not be forgotten, as well as the chain connectivity, which plays an essential role in the dynamics of these systems. The most evident signature of this ingredient is the sublinear increase in the mean squared atomic displacements ('Rouse-like', $\langle r^2(t) \rangle \propto t^{0.5}$) arising after the de-caging process (α -process) in contraposition to the linear

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regime (centre-of-mass diffusion, $\langle r^2(t) \rangle \propto t$) found in low-molecular-weight glass-formers. Another peculiarity of polymers is that, apart from librations or fast rotations of methyl groups, every motion, as local as it is, involves jumps over carbon–carbon rotational barriers and/or conformational changes in the chain. Due to chain connectivity and intramolecular barriers, each conformational transition induces a perturbation that propagates along the chain backbone and influences the neighbouring atoms within the same macromolecule. It is easy to imagine that these systems are dynamically very rich—they indeed exhibit an enormous number of internal degrees of freedom.

1,4-polybutadiene (1,4-PB) ($-\text{[CH}_2\text{—CH=CH—CH}_2\text{—]}_n$) has been considered as a ‘canonical’ glass-forming polymer due to its lack of side groups. A series of experimental works have been focused on understanding the dynamics of this system above its glass-transition temperature ($T_g \approx 180$ K), in particular in the so-called $\alpha\beta$ -merging region (in the neighbourhood of $T \approx 1.2T_g$). Such extensive studies have led to a puzzling situation: in the $\alpha\beta$ -region, apart from the structural relaxation and the Johari–Goldstein β -relaxation observed by dielectric spectroscopy [1], at least two other faster processes are found by neutron [1] and light [2, 3] scattering, both being slower than the fast microscopic process below 1–2 ps [4]. The controversy that is raised demands a molecular understanding of the dynamics in this ‘simple’ polymer, which can be facilitated by molecular dynamics (MD) simulations. With these ideas in mind, we performed fully atomistic MD simulations on this polymer. The obtained results clearly showed the occurrence of localized motions in the $\alpha\beta$ -regime, being active in the time window between the microscopic dynamics and the structural relaxation [5]. A recent molecular approach [5, 6] has identified such motions with the conformational transitions associated with the different secondary relaxations reported experimentally.

What is the impact of these motions on the application of the current theories for glass-forming dynamics to polymer data? Interestingly, we note that evidence of such local motions is found just in the time window where the mode coupling theory (MCT) for the glass transition [7] predicts the occurrence of *its* β -process, which *a priori* is not related to the Johari–Goldstein β -process. Then, can the MCT be applied to 1,4-PB?

In this work we present MD simulation results on 1,4-PB in the $\alpha\beta$ -region. We found a good agreement between simulation and experimental data by allowing for a small shift in the timescale. This could be interpreted as a difference in the glass transition temperature of about 15 K ($T_g^{\text{sim}} \approx T_g^{\text{exp}} - 15$ K). After a brief presentation of the outcome of the recently developed molecular approach [5, 6], we address the question of the applicability of the MCT to this polymer. A rather satisfactory check of the MCT predictions is achieved—at least formally—with a value of $T_c \approx 200$ K $\approx 1.2T_g^{\text{sim}}$ for the critical temperature, though with an unusually large value of the exponent parameter λ . After comparing this result with data from the literature on a variety of systems of different natures, we speculate on the origin of this observation as the result of two competing mechanisms for dynamic arrest: intermolecular packing and intramolecular barriers for conformational changes.

2. Molecular dynamics simulations

Fully atomistic MD simulations were carried out by using the Discover-3 module with the polymer consortium force-field under the Insight II environment from Accelrys. A cubic cell containing one polymer chain of 130 monomers was constructed at 280 K (about 100 K above the experimental T_g) by means of the amorphous cell protocol with periodic boundary conditions. The microstructure of the chain (40% *cis*; 53% *trans*; 7% vinyl units) was built to mimic that of the real sample. After equilibration at 280 K, the temperature was gradually lowered to the different investigated values (260, 240, 230, 220 and 200 K) by a series of

NPT dynamic steps at atmospheric pressure. For each temperature, once the equilibration density was reached, the energy of the structure was minimized and the system was dynamically equilibrated by a long run in the *NVT* ensemble. A subsequent *NVT* run was used to produce configurations for data analysis. Comparisons with extensive neutron scattering results showed that the simulated cell reproduces very fairly both structural and dynamical properties. A small shift in the simulation timescale provides an excellent agreement with experiments (simulation results are faster by less than half a decade). This shift roughly corresponds to a difference in temperature of 15 K. Further details can be found in [5, 6, 8].

3. Results

From the simulated atomic trajectories it is straightforward to calculate the self-part of the van Hove correlation function, i.e. the self-correlation function of the positions of a nucleus of type α at different times, $G_s^\alpha(r, t)$:

$$G_s^\alpha(r, t) = \frac{1}{N_\alpha} \left\langle \sum_{i=1}^{N_\alpha} \delta(r - |\vec{r}_i^\alpha(t) - \vec{r}_i^\alpha(0)|) \right\rangle. \quad (1)$$

Here r is the radial distance, \vec{r}_i^α is the position of atom i of type α , and N_α is the number of such nuclei in the sample. The brackets denote the ensemble average. Figures 1(a) and (b) show $G_s^H(r, t)$ (averaged for all the hydrogens in the cell) at the highest and lowest temperatures that were investigated. A qualitatively different behaviour can be found: at 280 K—well above T_g —the distribution continuously broadens and shifts towards larger distances with increasing time. These are typical signatures of a diffusive process such as that involved in the structural α -relaxation. In contrast, at 200 K the most prominent feature in the distribution function is the development of a second peak centred at about 3 Å. This double-peak structure reveals an underlying localized process as, for example, a jump between two positions separated by a distance $d \approx 2.5$ Å (roughly the distance between both peaks). An inspection of $G_s^\alpha(r, t)$ for each type of hydrogens in the sample (α : methyne hydrogen in *cis* [H_{1cis}] or in *trans* unit [H_{1trans}], or methylene hydrogen in *cis* [H_{2cis}] or *trans* unit [H_{2trans}]) reveals a strongly heterogeneous behaviour: the jumps occur in a distinct manner depending on the kind of atom considered. Finally, at this temperature, the characteristic features of an incipient diffusion can only be envisaged at the long-time limit of the simulation window.

The occurrence of these local motions strongly influences the short-time region of the correlation functions and other related quantities, such as, for example, the mean squared displacement of the hydrogens ($\langle r_H^2(t) \rangle$) (second moment of $G_s^H(r, t)$) (see figure 1(c)). Again, here a distinct behaviour is evident for high and low temperatures. At 280 K, the de-caging mechanism takes place almost immediately after the microscopic regime (picosecond region). In contrast, at 200 K the slope $\approx 1/2$ (the expected limit for the de-caging process towards the Rouse regime) is reached at rather long times (beyond 10 ns). Interestingly enough, in the region between the microscopic and the final de-caging dynamics, instead of the *a priori* expected plateau we find a progressive increase in $\langle r_H^2(t) \rangle$ (slope ≈ 0.2). This feature suggests an additional dynamical process occurring within the cage imposed by the neighbouring molecules. Such a process is just that observed as a local motion in $G_s^H(r, t)$ (see figures 1(a) and (b)). Moreover, at these low temperatures, such local processes also strongly affect the intermediate scattering function $F_s^\alpha(Q, t)$ (the Fourier transform of $G_s^\alpha(r, t)$ into the reciprocal space) in the time regime between the microscopic dynamics and the structural relaxation, as can be appreciated from figure 1(d) for the case of all hydrogens, $F_s^H(Q, t)$. A noticeable effect is observed even for low Q values [5], where intuitively diffusion would be the dominant relaxation mechanism.

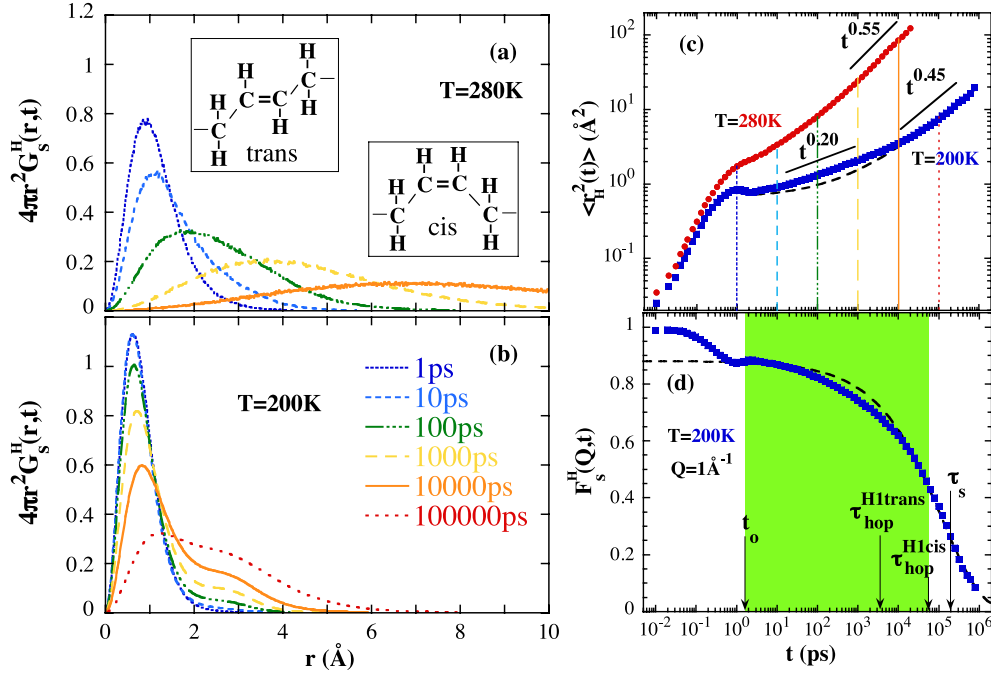


Figure 1. Simulations results for *all* the hydrogens. Panels (a) and (b): radial distribution functions (self) at 280 and 200 K, respectively. Times are given in (b). Insets in (a): scheme of the *cis* and *trans* units. Panel (c): $\langle r_H^2(t) \rangle$ at 280 K (circles) and 200 K (squares). The vertical lines indicate the times considered in (a) and (b). Panel (d): $F_s^H(Q, t)$ at 200 K and $Q = 1 \text{ \AA}^{-1}$. The thick dashed curves in (c) and (d) show the diffusive contribution from the jump-diffusion model (affected by the fast microscopic decay). Shaded area: time interval most influenced by the localized processes. Vertical arrows indicate the characteristic timescales for the microscopic dynamics (t_o), for the jumps identified for the methyne hydrogens in the *trans* ($\tau_{\text{hop}}^{\text{H1trans}}$) and *cis* ($\tau_{\text{hop}}^{\text{H1cis}}$) units (equation (2)), and for the structural relaxation (τ_s , KWW-time of the normalized dynamic structure factor at the first maximum of $S(Q)$, shown in figure 3(a)).

4. Molecular approach

As shown in recent works [5, 6], the phenomenology observed for the hydrogen motions in the $\alpha\beta$ -region of 1,4-PB can be well described by using a simple molecular approach which considers the simultaneous occurrence of localized motions and diffusion. In this framework, each atom jumps between two positions in an—in principle—assymmetric double-well potential, while it also participates in the anomalous diffusion related to the viscous flow associated with the structural relaxation. For a given kind of atom α , the hopping process is characterized by a jump distance d^α and a distribution of hopping times. This latter distribution leads to a relaxation function that we have modelled in a first approximation by a stretched exponential with a characteristic time τ_{hop}^α and a stretching exponent $\beta_{\text{hop}}^\alpha$. The sublinear or anomalous diffusion is represented by a diffusion coefficient D^α and a stretching exponent $\beta_{\text{diff}}^\alpha$. Finally, the asymmetry of the potential is depicted by the asymmetry energy ΔE^α . In terms of this model, the self-correlation function at $t > t_o \approx 1$ ps may be written as

$$G_s^\alpha(r, t) = (\eta_1^2 + \eta_2^2 + 2\eta_1\eta_2 \exp[-(t/\tau_{\text{hop}}^\alpha)^{\beta_{\text{hop}}^\alpha}]) G_s^{\alpha, \text{diff}}(r, t) + 2\eta_1\eta_2(1 - \exp[-(t/\tau_{\text{hop}}^\alpha)^{\beta_{\text{hop}}^\alpha}]) G_s^{\alpha, \text{diff}}(r - d^\alpha, t) \quad (2)$$

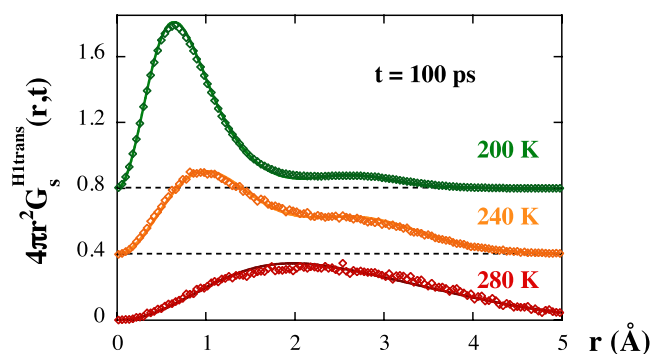


Figure 2. Radial self-correlation function for the double-bond hydrogens of the trans units at the different temperatures indicated and $t = 100$ ps. For clarity, the origins are shifted to the levels displayed by the horizontal dotted lines. The solid lines show the description obtained by the molecular model.

with the thermal occupation factors $\eta_1 = [1 + \exp(-\Delta E^\alpha/k_B T)]^{-1}$ and $\eta_2 = \eta_1 \exp(-\Delta E^\alpha/k_B T)$, and the self-correlation function for anomalous diffusion $G_s^{\alpha, \text{diff}}(r, t) = \exp(-r^2/[4(D^\alpha t)^{\beta_{\text{diff}}^\alpha} + \sigma_\alpha^2])/\tilde{V}$, where \tilde{V} is the appropriate normalization factor. Finally, in order to better model the initial distribution function at $t_o = 1$ ps, we use a sum of two initial Gaussians with slightly different widths, σ_α , which were kept constant for all fits at later times. This very simple model describes the simulation data very accurately [5, 6], as can be seen in figure 2 for the case of the methyne trans hydrogens at different temperatures. The systematic application of this model to the simulation data reveals the following results: (i) the temperature dependence of the diffusion coefficients D^α is consistent with the experimental expectation from viscosity measurements shifted by 15 K in the glass transition temperature value; (ii) the features deduced for this process (D^α , $\beta_{\text{diff}}^\alpha$) are independent of the type of hydrogen considered; (iii) the activation energy deduced for the jumps of the methyne hydrogens in the cis units is basically the same as that of the β -process monitored by dielectric spectroscopy [1] (we note that, consistently, the dipole moment in 1,4-PB is carried by the *cis* unit); (iv) the timescales obtained for the fastest hydrogens (methyne in the trans units) are in excellent agreement with those reported from Raman and Brillouin scattering [2, 3]. The jump distance involved in this case is well defined ($d^{\text{H1trans}} \approx 2.5$ Å) and could be attributed to counter-rotations of the trans units that leave the conformation of the rest of the chain practically unperturbed [9]. Other kinds of conformational transitions should be at the origin of the jump processes observed for the other hydrogens in the cell. This molecular approach thus provides a direct connection between the experimental observations of the different secondary relaxations reported for this polymer and the local motions involved in the conformational transitions taking place in the 1,4-PB building blocks.

We note that the region of maximum visibility of the localized motions is just located in the time window where one would expect the dynamics of the MCT β -process to be active. Thus, can the MCT be applied in this case and, if so, what is the result of a MCT analysis of our fully atomistic MD simulations on 1,4-PB?

5. Mode coupling theory approach

Since the introduction of the original MCT for hard-sphere systems [10], the theory has evolved to describe dynamic arrest of more complex molecules such as dumbbells [11], simplified models of water [12] and o-terphenyl [13], or ‘bead-chain’ models of polymers [14]. There are

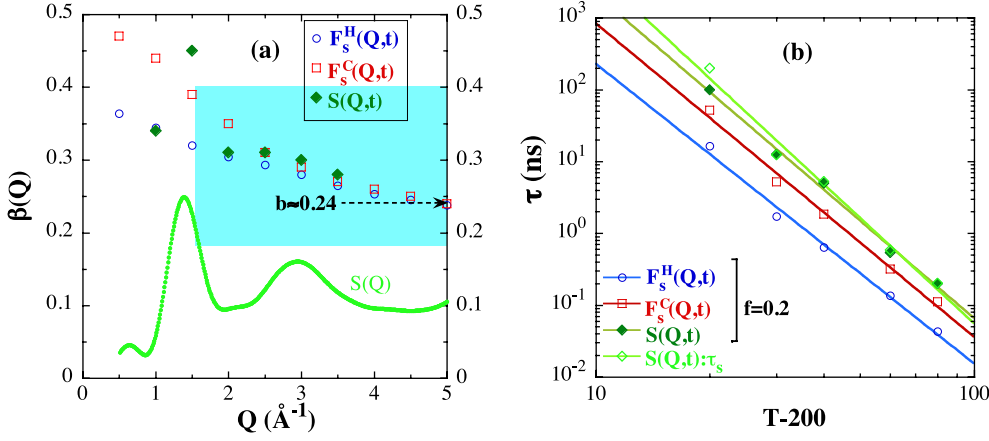


Figure 3. (a) Q -dependence of the stretching parameter β obtained for the intermediate scattering function of all hydrogens (circles), of all carbons (squares) and for the collective dynamic structure factor $S(Q, t)$ (diamonds). Shaded area: range deduced from neutron scattering on a fully protonated sample [16]. The static structure factor $S(Q)$ is shown for comparison in arbitrary units. (b) Temperature dependence of the characteristic timescales obtained from the same functions as in (a) defined as the time where the normalized correlation function takes the value of $f = 0.2$. The selected Q -value corresponds to the maximum of $S(Q)$. For $S(Q, t)$ the KWW time, τ_s (equation (4)), is also considered. Lines: fits to power laws with free exponents.

several, *a priori* universal, scaling and asymptotic laws for dynamic correlators and relaxation times (see below) which follow as mathematical consequences of the general structure of the MCT equations, and which are not restricted to monoatomic systems (see e.g. [11]). Since at present solutions of MCT equations are not available for atomistic models of real polymers such as 1.4-PB, simulation or experimental results cannot be compared directly with quantitative predictions for the quantities involved in the asymptotic laws mentioned. In such cases, a phenomenological analysis is usually performed, by relying on the validity of such predictions and deriving the quantities involved, such as fitting parameters, as we illustrate in the following.

As a first step towards an MCT analysis we may test the consistency of some of its predictions by considering the behaviour of parameters that can be easily obtained from a phenomenological data analysis. For example, the value of the von Schweidler exponent b can be deduced as the high- Q limit of the stretching parameter β [15],

$$\lim_{Q \rightarrow \infty} \beta(Q) = b \tag{3}$$

that results from the fit of the last stage of the decay of a given correlation function $\phi_Q(t)$ by a Kohlrausch–Williams–Watts (KWW) function

$$\phi_Q(t) = A \exp[-(t/\tau_\phi)^\beta]. \tag{4}$$

Figure 3(a) shows the results obtained for $\beta(Q)$ if we consider three different correlation functions $\phi_Q(t)$: the intermediate scattering functions calculated either for all hydrogens $F_s^H(Q, t)$ or all carbons $F_s^C(Q, t)$ in the system, and the dynamic structure factor $S(Q, t)$, which reflects all the possible atomic pair correlations in the system with identical weight. As can be seen in this figure, the high- Q limit of the KWW exponent is independent of the correlator considered and can be taken as $b \approx 0.24$. This value is consistent with the range reported in experiments on the real sample [16] ($\beta : 0.17 \dots 0.40$ for $1.5 \lesssim Q \lesssim 5 \text{\AA}^{-1}$). This

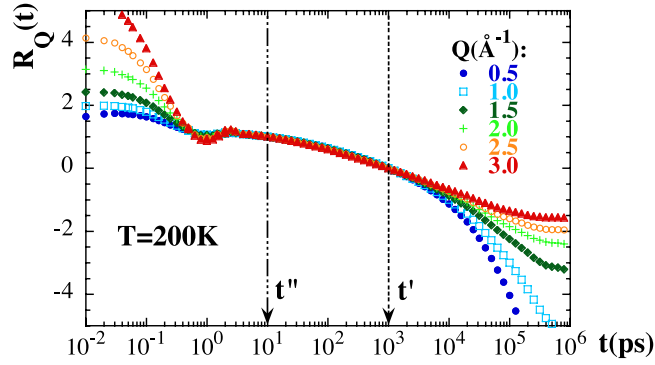


Figure 4. Test of the factorization theorem for $F_s^H(Q, t)$ of all hydrogens at 200 K.

b -value is significantly lower than those usually found for low-molecular-weight systems, as well as for simple bead-spring models for polymers (see below). Through the MCT relations

$$\lambda = \frac{\Gamma^2(1+b)}{\Gamma(1+2b)} = \frac{\Gamma^2(1-a)}{\Gamma(1-2a)}; \quad \gamma = \frac{1}{2a} + \frac{1}{2b} \quad (5)$$

this value of b implies unusually large values for λ and γ ($\lambda = 0.93$ and $\gamma = 4.9$).

Are these findings consistent with, for example, the T -dependence of the characteristic time for the structural relaxation τ ? In the MCT, the way this time approaches the critical temperature T_c is determined by the exponent γ as

$$\tau \propto |T - T_c|^{-\gamma}. \quad (6)$$

A priori, we do not know the value of T_c for our system. An estimation could be $T_c \approx 1.2T_g$, as is usually found in the literature. Taking into account the shift of the temperature in the MD simulations below that in the real sample ($\Delta T \approx 15$ K) and the experimental value of T_g for 1,4-PB ($T_g \approx 180$ K), we find $T_c \approx 200$ K. Figure 3(b) shows that the fit to power laws corresponding to equation (6) with $T_c = 200$ K deliver large exponents in the range $\gamma = 4.2 \dots 4.9$. As can be realized, these values are in very good agreement with the unusually large value of the γ -exponent independently deduced from equations (3) and (5). Thus, from this first simple analysis we can conclude that: (i) independent tests of MCT predictions are consistent; (ii) the exponent parameter λ seems to be unusually large (very close to 1).

Another prediction that can be tested without invoking fits is the factorization in the β -regime: $\phi_Q(t) = f_Q + h_Q G(t)$. If it holds, then the ratio

$$R_Q(t) = \frac{\phi_Q(t) - \phi_Q(t')}{\phi_Q(t'') - \phi_Q(t')} \quad (7)$$

must be Q -independent. As shown in figure 4 for $F_s^H(Q, t)$ at 200 K, the superposition of these functions is almost perfect in the MCT- β region ($2 \text{ ps} \lesssim t \lesssim 1 \text{ ns}$)—i.e. in the time window where the influence of the localized motions is most evident!

From now on we will thus consider that the MCT can be applied to our data with the critical exponents determined from the previous analysis. Being the β -scaling fulfilled, a description of the decay from the plateau by means of the MCT von Schweidler expansion

$$\phi_Q(t) = f_Q - H_1 Q t^b + H_2 Q t^{2b} + \dots \quad (8)$$

with a T -independent critical non-ergodicity parameter f_Q should be possible. We have first considered $F_s^H(Q, t)$ of all hydrogens at different temperatures. For each Q -value, we

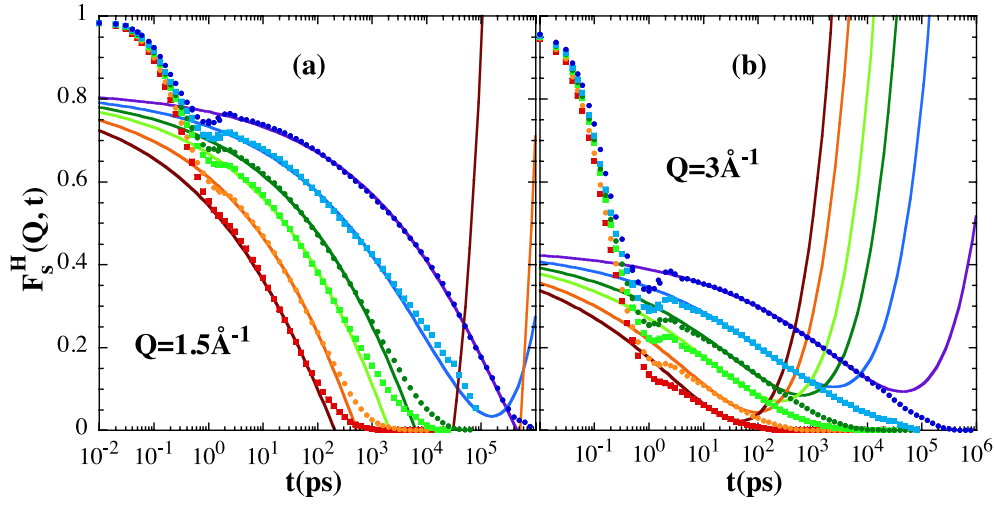


Figure 5. Fit of the MCT- β regime for $F_s^H(Q, t)$ of all hydrogens at $Q = 1.5 \text{ \AA}^{-1}$ (a) and $Q = 3 \text{ \AA}^{-1}$ (b) and 200, 220, 230, 240, 260 and 280 K (top to bottom).

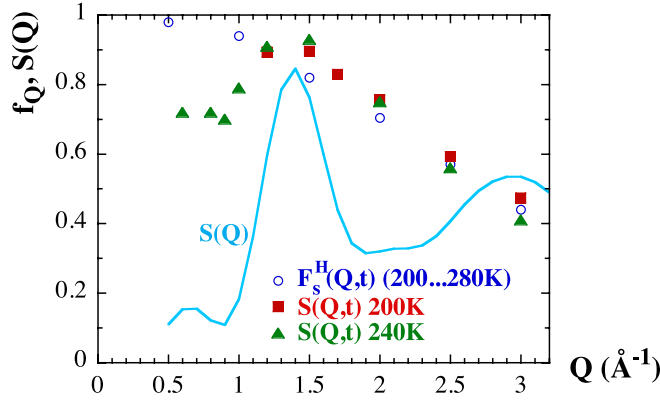


Figure 6. Q -dependence of the critical non-ergodicity parameter obtained from $F_s^H(Q, t)$ of all hydrogens (circles) and from $S(Q, t)/S(Q)$ at 200 K (squares) and 240 K (triangles). The static structure factor is shown for comparison in arbitrary units.

have described all the curves (figure 5) with a single value of f_Q . The resulting f_Q -values are displayed in figure 6, together with those obtained from the fit to equation (8) of the normalized collective dynamic structure factor $S(Q, t)/S(Q)$ at two temperatures. Again, for this correlator f_Q does not depend on temperature. Moreover, it is modulated by $S(Q)$. Both observations agree with MCT expectations.

Finally, we can obtain the value of the critical temperature T_c of our sample. This can be realized in two ways: (i) from the MCT description in the von Schweidler regime, since the H_{1Q} parameters in equation (8) must fulfil the relation

$$H_{1Q} \propto |T - T_c|^{\gamma b}; \quad (9)$$

(ii) from the timescale of the α -regime (equation (6)). Approach (i) leads to $T_c = 188 \text{ K}$ (see figure 7(a)), while through approach (ii) we arrive at $T_c = 198 \text{ K}$ (figure 7(b)). Both values, differing by only 5%, are compatible within error bars. It is worthy of remark that the

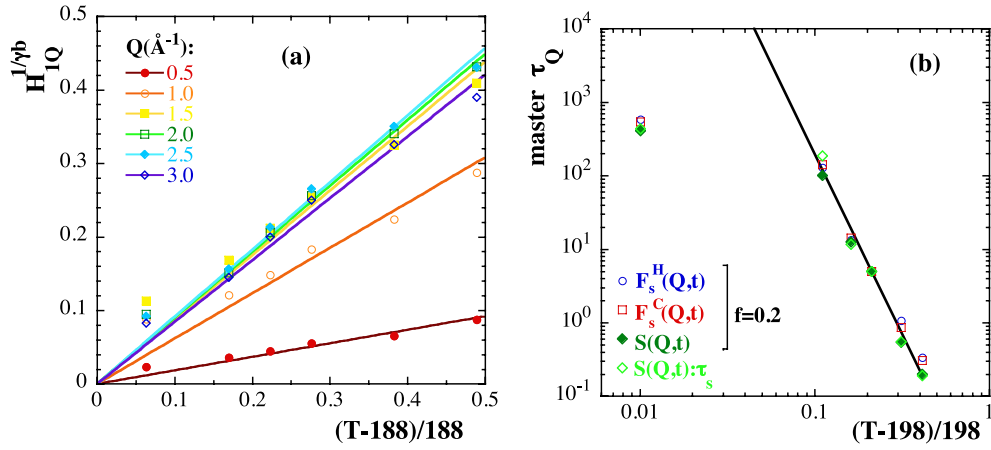


Figure 7. Tests of the MCT predictions. Panel (a): β -scaling (equation (9) applied to $F_s^H(Q, t)$); panel (b) α -relaxation times defined as in figure 3(b) and scaled into a master curve (equation (6)). As is usually observed, deviations from MCT predictions, which are attributed to ill-defined ergodicity-restoring processes, occur at temperatures very close to T_c .

Table 1. Values of the MCT critical exponents for different systems.

System	a	b	γ	λ	Reference
Hard spheres	0.31	0.58	2.5	0.74	[18]
<i>o</i> -terphenyl	0.30	0.54	2.6	0.76	[19]
Silica	0.32	0.62	2.3	0.71	[20]
Water	0.29	0.51	2.7	0.78	[21]
Bead-spring polymer (MD)	0.35	0.75	2.1	0.63	[22]
Bead-spring polymer (MCT)	0.32	0.60	2.4	0.72	[22]
Polyethylene (united atom)	0.27	0.46	2.9	0.81	[23]
Polybutadiene (united atom)	0.21	0.30	4.1	0.90	[24]
Polybutadiene (fully atomistic)	0.18	0.24	4.9	0.93	This work

range of values obtained for T_c is also compatible with the T_c value reported from experimental studies on this polymer ($T_c \approx 214$ K) [17], taking into account the above-mentioned shift in temperature between the simulated and real samples.

6. Discussion

We have shown that atomic motions in the $\alpha\beta$ -regime ($200 \text{ K} \leq T \leq 280 \text{ K}$) of 1,4-PB consist of localized and diffusive contributions. From a molecular approach, the localized processes can be associated with conformational transitions involving the rigid building blocks of 1,4-PB that are responsible for the experimentally observed secondary relaxations (including the Johari–Goldstein β -relaxation). These motions strongly affect the dynamics between the microscopic region and the structural relaxation, i.e. in the MCT β -regime dominated by the ‘cage’ effect imposed by the surrounding neighbours. However, at least formally, MCT can be applied to 1,4-PB results and its main predictions are fulfilled. Apparently, the MCT analysis reflects the presence of the local conformational changes through the large value deduced for the exponent parameter λ . As mentioned above and shown in table 1, MCT analysis of different systems [18–21] yields much smaller λ -values. Low-molecular-weight systems, such as

o-terphenyl, silica or water, show values similar to that of hard spheres. However, adding the ingredient of chain connectivity in bead-spring models does not yield significant changes in λ . We note that bead-spring models referred to in [22] deal with *fully flexible* chains and do not include rotational barriers. A significant change is induced by incorporating these barriers, as in a united atom model of polyethylene [23]. Even larger λ -values are obtained for 1,4-PB. Seemingly, the presence of double bonds along the main chain of 1,4-PB allows for counter-rotations leading to particular local motions of parts of the macromolecule which are not found in the most simple polyethylene. Our study suggests that the role of the intramolecular barriers is crucial in determining the dynamics within the cage and, as a consequence, the observed large value of λ . A large value of λ (≈ 0.9) is also reported for the united atom model of 1,4-PB [24]. However, in that work the real-space correlation functions are not analysed and the presence of localized motions is not considered. Nevertheless, it is worth remarking that the localized motions identified in the present work do not correspond to relative motions of the hydrogens with respect to the carbons, but to local conformational transitions also involving backbone atoms. Thereby these localized motions certainly have to be present also in a united atom model of 1,4-PB.

Values of λ approaching 1 point to a situation close to an MCT higher-order transition ($\lambda = 1$) [25]. Features associated with such transitions have recently been reported for attractive colloids [26] and a series of systems showing strong confinement effects [27]. The origin of the observed anomalous relaxation features is attributed to the presence of several competing mechanisms for dynamic arrest—steric repulsion and reversible bond formation in [26], bulk-like caging and confinement in [27]. In the case of the $\alpha\beta$ -region of real polymers, we may speculate that there also exist two active competing mechanisms leading to dynamic arrest: (i) packing, of intermolecular character and present in all glass-forming systems; (ii) barriers for conformational changes (β -like), of intramolecular origin, which are specific of macromolecular systems.

Acknowledgments

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References

- [1] Arbe A, Richter D, Colmenero J and Farago B 1996 *Phys. Rev. E* **54** 3853
- [2] Aouadi A, Lebon M J, Dreyfus C, Strube B, Steffen W, Patkowski A and Pick R M 1997 *J. Phys.: Condens. Matter* **9** 3803
- [3] Ding Y, Novikov V N and Sokolov A P 2004 *J. Polym. Sci. B* **42** 994
- [4] Zorn R, Arbe A, Colmenero J, Frick B, Richter D and Buchenau U 1995 *Phys. Rev. E* **52** 781
- [5] Colmenero J, Arbe A, Alvarez F, Narros A, Monkenbusch M and Richter D 2005 *Europhys. Lett.* **71** 262
- [6] Colmenero J, Arbe A, Alvarez F, Narros A, Monkenbusch M and Richter D 2007 *J. Chem. Phys.* submitted
- [7] Götze W 1991 *Liquids, Freezing, Glass Transition* ed J P Hansen, D Levesque and J Zinn-Justin (Amsterdam: North-Holland) p 287
- [8] Narros A, Arbe A, Alvarez F, Colmenero J, Zorn R, Schweika W and Richter D 2005 *Macromolecules* **38** 9847
- [9] Kim E G and Mattice W M 2002 *J. Chem. Phys.* **117** 2389
- [10] Bengtzelius U, Götze W and Sjölander A 1984 *J. Phys. C: Solid State Phys.* **17** 5915
- [11] Chong S H and Götze W 2002 *Phys. Rev. E* **65** 051201
- [12] Theis C, Sciortino F, Latz A, Schilling R and Tartaglia P 2000 *Phys. Rev.* **62** 1856
- [13] Chong S H and Sciortino F 2004 *Phys. Rev. E* **69** 051202

- [14] Chong S H and Fuchs M 2002 *Phys. Rev. Lett.* **88** 185702
- [15] Fuchs M 1994 *J. Non-Cryst. Solids* **172** 241
- [16] Zorn R, Richter D, Frick B and Farago B 1993 *Physica A* **201** 52
- [17] Frick B, Farago B and Richter D 1990 *Phys. Rev. Lett.* **64** 2921
- [18] Franosch T, Fuchs M, Götze W, Mayr M R and Singh A P 1997 *Phys. Rev. E* **55** 7153
- [19] Mossa S, Di Leonardo R, Ruocco G and Sampoli M 2000 *Phys. Rev. E* **62** 612
- [20] Horbach J and Kob W 2001 *Phys. Rev. E* **64** 041503
- [21] Sciortino F, Fabbian L, Chen S-H and Tartaglia P 1997 *Phys. Rev. E* **56** 5397
- [22] Baschnagel J and Varnik F 2005 *J. Phys.: Condens. Matter* **17** R851
- [23] Van Zon A and De Leeuw S W 1999 *Phys. Rev. E* **60** 6942
- [24] Paul W, Bedrov D and Smith G D 2006 *Phys. Rev. E* **74** 021501
- [25] Götze W and Haussmann R 1988 *Z. Phys. B* **72** 403
Sperl M 2003 *Phys. Rev. E* **68** 031405
Krakoviack V 2005 *Phys. Rev. Lett.* **94** 065703
- [26] Sciortino F, Tartaglia P and Zaccarelli E 2003 *Phys. Rev. Lett.* **91** 268301
- [27] Moreno A J and Colmenero J 2006 *J. Chem. Phys.* **124** 184906
Moreno A J and Colmenero J 2006 *Phys. Rev. E* **74** 021409
Moreno A J and Colmenero J 2006 *J. Chem. Phys.* **125** 164507